

This is G o o g l e's cache of <http://wbiomed.curtin.edu.au/teach/biochem/tutorials/AAs/AA.html> as retrieved on Jun 16, 2006 02:59:24 GMT.

G o o g l e's cache is the snapshot that we took of the page as we crawled the web.

The page may have changed since that time. Click here for the [current page](#) without highlighting.

This cached page may reference images which are no longer available. Click here for the [cached text only](#).

To link to or bookmark this page, use the following url: <http://www.google.com/search?q=cache:ucu-enNz8F4J:wbiomed.curtin.edu.au/teach/biochem/tutorials/AAs/AA.html+alpha+amino+acid&hl=en&gl=us&ct=clnk&cd=3>

*Google is neither affiliated with the authors of this page nor responsible for its content.*

These search terms have been highlighted: **alpha amino acid**

## A Review of Amino Acids



This brief tutorial covers the chemical and structural properties of the 20 **Amino** Acids commonly found in proteins.

The tutorial contains a number of graphic images, it may take a while to download, so please be patient. After reviewing the **amino** acids, try out the [Interactive Quiz](#). 😊

**Note:** The quiz requires a browser that recognises Frames and JavaScript, for example - Netscape Navigator 2.0 or above.

## Amino Acid Properties

**Amino** acids are the basic structural units of proteins. An **alpha-amino acid** consists of an **amino** group, a carboxyl group, a hydrogen atom, and a distinctive R group bonded to a carbon atom, which is called the **alpha-carbon** because it is adjacent to the carboxyl (acidic) group. An R group is referred to as a *side chain*. (Stryer, 1988)






**Amino** Acids are commonly classified into the following groups based on the chemical and/or structural properties of their *side chains* :

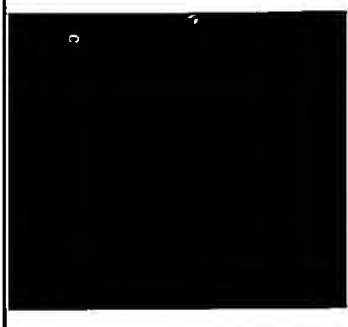
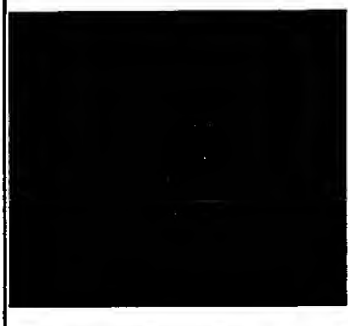





- [Aliphatic Amino Acids](#)
- [A Cyclic Amino Acid](#)
- [AAs with Hydroxyl or Sulfur-containing side chains](#)
- [Aromatic Amino Acids](#)
- [Basic Amino Acids](#)
- [Acidic Amino Acids and their Amides](#)

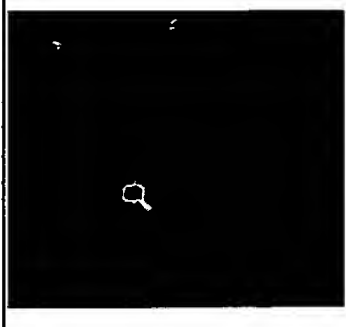
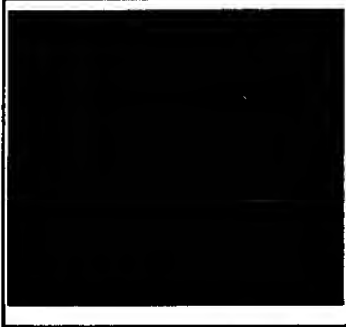




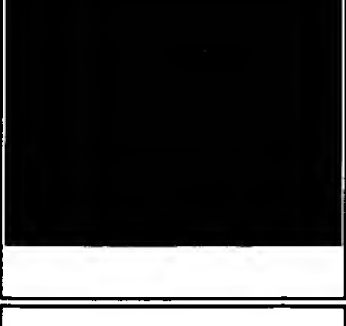
Amino Acid Structures

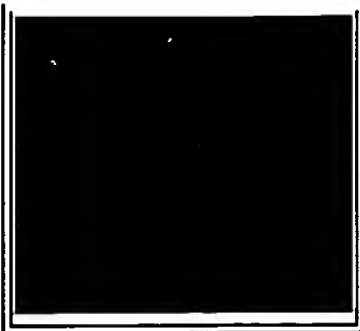
To view the **amino acid** structures using [Rasmol](#) click on the appropriate **amino acid** images below. A script is available ([aacolors](#)) to color the molecules as per the image.

When you are finished viewing the **amino acid**, type  
    **zap** (enter)  
in the command line window to close the current molecule before selecting another **amino acid** for viewing. Quit the Rasmol application by typing  
    **exit** (enter)  
in the command line window. Consult the online [Rasmol Manual](#) if you need further help using Rasmol.

Rasmol Image	Amino Acid	3-letter code	1-letter code	Properties	Structure (un-ionised form)
	Alanine	Ala	A	aliphatic hydrophobic neutral	$\begin{array}{c} \text{O} \\ \parallel \\ \text{H}_2\text{N}-\text{CH}-\text{C}-\text{OH} \\   \\ \text{CH}_3 \end{array}$
	Arginine	Arg	R	polar hydrophilic charged (+)	$\begin{array}{c} \text{O} \\ \parallel \\ \text{H}_2\text{N}-\text{CH}-\text{C}-\text{OH} \\   \\ \text{CH}_2 \\   \\ \text{CH}_2 \\   \\ \text{CH}_2 \\   \\ \text{NH} \\   \\ \text{C}=\text{NH} \\   \\ \text{NH}_2 \end{array}$
	Asparagine	Asn	N	polar hydrophilic neutral	$\begin{array}{c} \text{O} \\ \parallel \\ \text{H}_2\text{N}-\text{CH}-\text{C}-\text{OH} \\   \\ \text{CH}_2 \\   \\ \text{C}=\text{O} \\   \\ \text{NH}_2 \end{array}$
	Aspartate	Asp	D	polar hydrophilic charged (-)	$\begin{array}{c} \text{O} \\ \parallel \\ \text{H}_2\text{N}-\text{CH}-\text{C}-\text{OH} \\   \\ \text{CH}_2 \\   \\ \text{C}=\text{O} \\   \\ \text{OH} \end{array}$
	Cysteine	Cys	C	polar hydrophobic neutral	$\begin{array}{c} \text{O} \\ \parallel \\ \text{H}_2\text{N}-\text{CH}-\text{C}-\text{OH} \\   \\ \text{CH}_2 \\   \\ \text{SH} \end{array}$

	Glutamine	Gln	Q	polar hydrophilic neutral	$\begin{array}{c} \text{O} \\ \parallel \\ \text{H}_2\text{N}-\text{CH}-\text{C}-\text{OH} \\   \\ \text{CH}_2 \\   \\ \text{CH}_2 \\   \\ \text{C}=\text{O} \\   \\ \text{NH}_2 \end{array}$
	Glutamate	Glu	E	polar hydrophilic charged (-)	$\begin{array}{c} \text{O} \\ \parallel \\ \text{H}_2\text{N}-\text{CH}-\text{C}-\text{OH} \\   \\ \text{CH}_2 \\   \\ \text{CH}_2 \\   \\ \text{C}=\text{O} \\   \\ \text{OH} \end{array}$
	Glycine	Gly	G	aliphatic neutral	$\begin{array}{c} \text{O} \\ \parallel \\ \text{H}_2\text{N}-\text{CH}-\text{C}-\text{OH} \\   \\ \text{H} \end{array}$
	Histidine	His	H	aromatic polar hydrophilic charged (+)	$\begin{array}{c} \text{O} \\ \parallel \\ \text{H}_2\text{N}-\text{CH}-\text{C}-\text{OH} \\   \\ \text{CH}_2 \\   \\ \text{C}_5\text{H}_4\text{N} \end{array}$
	Isoleucine	Ile	I	aliphatic hydrophobic neutral	$\begin{array}{c} \text{O} \\ \parallel \\ \text{H}_2\text{N}-\text{CH}-\text{C}-\text{OH} \\   \\ \text{CH}-\text{CH}_3 \\   \\ \text{CH}_2 \\   \\ \text{CH}_3 \end{array}$
	Leucine	Leu	L	aliphatic hydrophobic neutral	$\begin{array}{c} \text{O} \\ \parallel \\ \text{H}_2\text{N}-\text{CH}-\text{C}-\text{OH} \\   \\ \text{CH}_2 \\   \\ \text{CH}-\text{CH}_3 \\   \\ \text{CH}_3 \end{array}$
	Lysine	Lys	K	polar hydrophilic charged (+)	$\begin{array}{c} \text{O} \\ \parallel \\ \text{H}_2\text{N}-\text{CH}-\text{C}-\text{OH} \\   \\ \text{CH}_2 \\   \\ \text{CH}_2 \\   \\ \text{CH}_2 \\   \\ \text{CH}_2 \\   \\ \text{NH}_2 \end{array}$

	Methionine	Met	M	hydrophobic neutral	$\begin{array}{c} \text{H}_2\text{N}-\text{CH}-\text{C}(=\text{O})-\text{OH} \\   \\ \text{CH}_2 \\   \\ \text{CH}_2 \\   \\ \text{S} \\   \\ \text{CH}_3 \end{array}$
	Phenylalanine	Phe	F	aromatic hydrophobic neutral	$\begin{array}{c} \text{H}_2\text{N}-\text{CH}-\text{C}(=\text{O})-\text{OH} \\   \\ \text{CH}_2 \\   \\ \text{C}_6\text{H}_5 \end{array}$
	Proline	Pro	P	hydrophobic neutral	$\begin{array}{c} \text{O} \\    \\ \text{C}-\text{OH} \\   \\ \text{HN} \end{array}$
	Serine	Ser	S	polar hydrophilic neutral	$\begin{array}{c} \text{H}_2\text{N}-\text{CH}-\text{C}(=\text{O})-\text{OH} \\   \\ \text{CH}_2 \\   \\ \text{OH} \end{array}$
	Threonine	Thr	T	polar hydrophilic neutral	$\begin{array}{c} \text{H}_2\text{N}-\text{CH}-\text{C}(=\text{O})-\text{OH} \\   \\ \text{CH}-\text{OH} \\   \\ \text{CH}_3 \end{array}$
	Tryptophan	Trp	W	aromatic hydrophobic neutral	$\begin{array}{c} \text{H}_2\text{N}-\text{CH}-\text{C}(=\text{O})-\text{OH} \\   \\ \text{CH}_2 \\   \\ \text{Indole ring} \end{array}$
	Tyrosine	Tyr	Y	aromatic polar hydrophobic	$\begin{array}{c} \text{H}_2\text{N}-\text{CH}-\text{C}(=\text{O})-\text{OH} \\   \\ \text{CH}_2 \\   \\ \text{C}_6\text{H}_4 \\   \\ \text{OH} \end{array}$

	Valine	Val	V	aliphatic hydrophobic neutral	$\begin{array}{c} \text{H}_2\text{N}-\text{CH}-\overset{\overset{\text{O}}{\parallel}}{\text{C}}-\text{OH} \\   \\ \text{CH}-\text{CH}_3 \\   \\ \text{CH}_3 \end{array}$
--	--------	-----	---	-------------------------------------	---

<a href="#">Home</a>	<a href="#">Lectures</a>	<a href="#">Tutorials</a>	<a href="#">Gallery</a>	<a href="#">Topics</a>	<a href="#">Help</a>
----------------------	--------------------------	---------------------------	-------------------------	------------------------	----------------------

This is **G o o g l e**'s cache of <http://v3.espacenet.com/textdes?DB=EPODOC&IDX=CN1720025&F=0&QPN=CN1720025> as retrieved on Jun 5, 2006 18:21:43 GMT.

**G o o g l e**'s cache is the snapshot that we took of the page as we crawled the web.

The page may have changed since that time. [Click here for the current page without highlighting.](#)

This cached page may reference images which are no longer available. [Click here for the cached text only.](#)

To link to or bookmark this page, use the following url: [http://www.google.com/search?](http://www.google.com/search?q=cache:V2jJwAO6wjAJ:v3.espacenet.com/textdes%3FDB%3DEPODOC%26IDX%3DCN1720025%26F%3D0%26QPN%3DCN1720025+lactam+gabapentin&hl=en&gl=us&ct=clnk&cd=13)

[q=cache:V2jJwAO6wjAJ:v3.espacenet.com/textdes%3FDB%3DEPODOC%26IDX%3DCN1720025%26F%3D0%26QPN%3DCN1720025+lactam+gabapentin&hl=en&gl=us&ct=clnk&cd=13](http://www.google.com/search?q=cache:V2jJwAO6wjAJ:v3.espacenet.com/textdes%3FDB%3DEPODOC%26IDX%3DCN1720025%26F%3D0%26QPN%3DCN1720025+lactam+gabapentin&hl=en&gl=us&ct=clnk&cd=13)

*Google is neither affiliated with the authors of this page nor responsible for its content.*

These search terms have been highlighted: **lactam gabapentin**

## Gabapentin tablets and methods for their preparation

Description of corresponding document: **US2006039968**

### FIELD OF THE INVENTION

[0001] The invention is generally directed to stable **gabapentin** tablets prepared by wet granulation.

### BACKGROUND OF THE INVENTION

[0002] **Gabapentin** is an anti-epileptic drug indicated as adjunctive therapy in the treatment of partial seizures with and without secondary generalization in adults with epilepsy. **Gabapentin** exists in a crystalline form and exhibits poor compressibility and compactibility. These detrimental characteristics of **gabapentin** cause capping and lamination defects during compression of **gabapentin** into tablets.

[0003] Conventionally, these problems are overcome by incorporating compression aids in the formulation. However, the more excipients, such as compression aids, that are used in a composition the more expensive and time-consuming commercial production becomes. Moreover, increasing the amount of excipient increases the size of the tablet, which can result in overly large tablets, an undesirable result for pediatric use or for those patients who have difficulty in swallowing.

[0004] Moreover, including a large amount and/or number of excipients in a **gabapentin** formulation results in stability problems, such as degradation. For example, **gabapentin** has been found to degrade into **lactam**, resulting in a decrease in the potency of **gabapentin** over time. Because of the decrease in potency, it is necessary to avoid degradation of **gabapentin** over the shelf life of the product. Generally, shelf life of the product is two years from completion of manufacture. The level of degradation over the shelf life of the tablets can be determined by storing the product in closed containers for a three-month period at 40[deg.] C. and 75% relative humidity. It is generally accepted that tablets containing **gabapentin** should have no more than about 0.4% by weight of **lactam**, as determined by High Performance Liquid Chromatography (HPLC), at the end of this three-month period.

[0005] To combat the **lactam** formation and provide product stability, U.S. Pat. No. 6,054,482 discloses the importance of (a) starting with **gabapentin** raw material that contains 0.5% or less of corresponding **lactam**; (b) not allowing the anion of a mineral acid in the composition to exceed 20 ppm, and (c) using a specifically selected adjuvant that is not adverse to **gabapentin** stability.

[0006] In an attempt to achieve this, the patent discloses a method that includes hydrolyzing **gabapentin** with a semi-concentrated mineral acid and then converting **gabapentin** into solid pharmaceutical compositions containing hydroxypropylmethylcellulose (HPMC), polyvinylpyrrolidone, crospovidone, maize starch, cyclodextrin, talcum, copolymer of dimethylaminomethacrylic acid and/or neutral methacrylic acid ester.

[0007] Another difficulty encountered in producing **gabapentin** tablets is that **gabapentin** is not amenable to traditional wet granulation techniques. Because the viscosity of the binder solution increases with the possible necessary increase in binder content, to apply a functional amount of binder for **gabapentin** the amount of solvent has to be increased. Increasing the amount of solvent, however, results in a wet granulation that is in a semi-liquid state and is not suitable for conventional drying methods. Therefore, to avoid the semi-liquid state, the wet granulation technique must be done in multiple stages in which a portion of binder solution is added, followed by drying, then the next portion of binder solution is added, and so forth. This becomes a time consuming and expensive process.

[0008] Purepac, the assignee of U.S. Pat. No. 6,294,198, has addressed this problem in the patent by using a spray-coating method in which a binder is dissolved in a solvent to form a binder solution that is then spray-coated on the drug particles. By using this method, substantially all of the solvent is evaporated as it is applied, leaving a film of binder around the drug particles. This process is conducted at or below room temperature.

### SUMMARY OF THE INVENTION

[0009] In one general aspect there is provided a wet granulation method for preparing stable **gabapentin** tablets. The wet granulation method includes: forming a mixture by dry mixing of a first portion of a binder with the **gabapentin**, one or more excipients, or a combination of the **gabapentin** and the one or more excipients; and adding a second portion of the binder to the mixture, wherein the second portion of the binder is in the form of a solution or dispersion.

[0010] Embodiments of the wet granulation method may include one or more of the following features. For example, the method may further include one or more of mixing the second portion of the binder with the mixture to form granules, drying the granules, mixing one or more excipients with the granules, and compressing into tablets.

[0011] The tablets may have a **lactam** content less than 0.1% by weight of **gabapentin** after one month of storage at 40[deg.] C. and 75% humidity, less than 0.2% by weight of **gabapentin** after two months of storage at 40[deg.] C. and 75% humidity, or less than 0.4% by weight of **gabapentin** after three months of storage at 40[deg.] C. and 75% humidity. In particular, the tablets may have a **lactam** content less than about 0.2% by weight of **gabapentin** after three months of storage at 40[deg.] C. and 75% humidity

[0012] The binder solution or dispersion may be prepared in water alone or in a mixture of water with one or more of ethanol, isopropyl alcohol, and acetone. The binder solution or dispersion may be prepared in water. The binder solution or dispersion may be prepared in a mixture of water and ethanol. The ratio of drug to binder may be between about 1:0.01 and about 1:1. The binder may be one or more of hydroxypropyl cellulose, hydroxypropyl methylcellulose, polyvinylpyrrolidone, copolyvidone, and sugars. In particular, the binder may be hydroxypropyl cellulose and/or copolyvidone.

[0013] The **gabapentin** may be the free base hydrated form, a monohydrate, or other pharmaceutically acceptable salt thereof. The **gabapentin** may have an anion of the mineral acid at about 100 ppm or less as calculated by chloride content. In particular, the anion of the mineral acid may be between about 20 and about 100 ppm.

[0014] The excipients mixed with the **gabapentin** or the granules may be one or more of disintegrants, fillers, stabilizers, lubricants, colorants, flavors, and glidants.

[0015] The disintegrant may be one or more of microcrystalline cellulose, sodium starch glycolate, crosslinked carboxy methylcellulose, and croscopovidone. The disintegrant may be between about 0.5% w/w to about 15% w/w of the tablet.

[0016] The filler may be one or more of lactose, microcrystalline cellulose, mannitol, and dicalcium phosphate.

[0017] The stabilizer may be one or more of poloxamer, cremophor, anionic surfactants, cationic surfactants, and nonionic surfactants. The stabilizer may be about 0.1% w/w to about 10% w/w of the tablet.

[0018] The lubricant may be one or more of magnesium stearate, stearic acid, and stearyl fumarate.

[0019] The wet granulation method may further include coating the tablet. The coating may be one or more of a hydrophilic polymer, hydroxypropyl cellulose, hydroxypropyl methylcellulose, polyvinyl pyrrolidone, and polyvinyl alcohol. The coated tablet may have a friability of less than 1% w/w and an initial friability of less than about 0.1% w/w. The coated tablet may have a hardness of about 10 Kp to 30 Kp, and an initial hardness of between about 20 Kp and about 25 Kp.

[0020] In another general aspect there is provided a **gabapentin** tablet formed by wet granulation. The **gabapentin** tablet has a **lactam** content of less than 0.4% by weight of **gabapentin** after three months of storage at 40[deg.] C. and 75% humidity.

[0021] Embodiments of the tablet may include one or more of the features described above or following. For example, the wet granulation may include forming a mixture by dry mixing of a first portion of a binder with the **gabapentin**, one or more excipients, or a combination of the **gabapentin** and the one or more excipients; and adding a second portion of the binder to the mixture, wherein the second portion of the binder is in the form of a solution or dispersion.

[0022] In another general aspect there is provided a method of one or more of treating epilepsy, treating neuropathic pain, anticonvulsant therapy, treating post poliomyelitis pain, treating amyotrophic lateral sclerosis, controlling rapid cycling and mixed bipolar states, treating the pain of diabetic neuropathy, and as a prophylactic agent for patients with migraine headaches, the method including providing a **gabapentin** tablet prepared by wet granulation.

[0023] Embodiments of the tablet may include one or more of the features described above or following. For example, the wet granulation may include forming a mixture by dry mixing of a first portion of a binder with the **gabapentin**, one or more excipients, or a combination of the **gabapentin** and the one or more excipients; and adding a second portion of the binder to the mixture, wherein the second portion of the binder is in the form of a solution or dispersion. The **gabapentin** tablets may have a **lactam** content of less than 0.4% by weight of **gabapentin** after three months of storage at 40[deg.] C. and 75% humidity.

[0024] The details of one or more embodiments of the inventions are set forth in the description below. Other features, objects and advantages of the inventions will be apparent from the description and claims.

#### DETAILED DESCRIPTION OF THE INVENTION

[0025] We have now discovered that stable **gabapentin** tablets can be prepared by a wet granulation method and, unlike the disclosure in U.S. Pat. No. 6,054,482, without having to limit use of **gabapentin** to a **gabapentin** having an anion of a mineral acid (calculated as chloride content) less than 20 ppm. The resulting tablets are not only free from capping and lamination defects but also have better hardness and are stable.

[0026] In one embodiment, the present invention relates to a wet granulation method for preparing stable **gabapentin** tablets, in which the tablets after three months of storage at 40[deg.] C. and 75% humidity have a **lactam** content less than 0.4% by weight of **gabapentin**.

[0027] This stability is provided by using a wet granulation method which includes dry mixing of a part of the binder with the drug, other excipients, or both; and then adding the rest of the binder in the form of a solution/dispersion. The addition of the binder in two portions is advantageous. First, the quantity of solvent used for preparing the binder solution is reduced to a minimum, which makes it possible to add binder solution in a single step. The two-portion



addition also reduces the duration of exposure of **gabapentin** to the solvent, which can further reduce the likelihood of polymorph conversion and/or changes in crystal structure in **gabapentin**. Second, since the use of solvent is kept to a minimum, there is an improvement in the safety and environmental impact of the process.

[0028] Furthermore, the wet granulation method described herein also may be applied to other active drugs, and, in particular, those that have poor compressibility and compactibility.

[0029] In applying the method to **gabapentin**, any binder that is compatible with **gabapentin** may be used. For example, the binder may be selected from hydroxypropyl cellulose, hydroxypropylmethyl cellulose, polyvinyl pyrrolidone, copolyvidone, sugars, or a combination thereof. The binder may be dissolved or dispersed in a solvent such as water alone or a mixture of water and ethanol, isopropyl, alcohol and/or acetone. The concentration of binder in the solution will depend upon the components used and the desired viscosity. Typically, the drug to binder ratio will vary from about 1:0.01 to about 1:1. The binder solution/dispersion may be prepared by any method that permits dissolution of binder to produce a homogenous solution, mixture or dispersion, such that formulations may be prepared that will contain a uniform amount of the binder.

[0030] The **gabapentin** may be present as a free base, hydrated form, such as monohydrate, or any other pharmaceutically acceptable salt thereof. The amount of an anion of the mineral acid (calculated as chloride content) may vary up to about 100 ppm.

[0031] The other excipients in the formulation may be selected from one or more of disintegrants, fillers, stabilizers, lubricants, colorants, flavors and glidants.

[0032] The disintegrant may be one or more of microcrystalline cellulose, sodium starch glycolate, crosslinked carboxy methylcellulose, crospovidone, other suitable disintegrants, or a combination thereof. The disintegrant may be present intragranularly, as well as extragranularly. The disintegrant may be used at a concentration of about 0.5% w/w to about 15% w/w of the tablet.

[0033] The fillers may be one or more of any conventional filler, such as lactose, microcrystalline cellulose, mannitol, dicalcium phosphate, other suitable fillers, or a combination thereof.

[0034] The stabilizer may be one or more of poloxamer, cremophor, other anionic, cationic, nonionic surfactants, or a combination thereof. The stabilizer may be used in concentration of between about 0.1% w/w to about 10% w/w of tablet.

[0035] The lubricant may be one or more of magnesium stearate, stearic acid, sodium stearyl fumarate, other suitable lubricants, or combinations thereof.

[0036] The method may be carried out using the following steps:

(i) **Gabapentin** is mixed with one or more disintegrants in a mixer.

(ii) The binder is divided into two portions, one portion is mixed with the **gabapentin**-disintegrant mixture and the remaining portion is dissolved in a sufficient quantity of granulating solvent to prepare a binder solution.

(iii) The binder solution is then mixed with the **gabapentin**-disintegrant-binder mixture of step (ii) in a low shear mixer.

(iv) The granules of step (iii) are dried in a fluidized bed dryer.

(v) The dried granules are mixed with rest of the excipients, e.g., stabilizers, fillers, glidants, disintegrants (extragranular) and lubricants and compressed into tablets using appropriate tooling.

[0042] For ease of swallowing and to enhance the aesthetic appeal, it may be desirable to coat the tablet as an optional step. The coating may be made of one or more hydrophilic polymers such as hydroxypropyl cellulose, hydroxypropyl methylcellulose, polyvinyl pyrrolidone and polyvinyl alcohol.

[0043] The tablets prepared by the above method have a hardness of about 10 Kp to about 30 Kp and a friability of less than 1% w/w.

[0044] The **lactam** content of the **gabapentin** tablet made by the above method does not exceed 0.4% by weight of **gabapentin** after storage for three months at 40[deg.] C. and 75% relative humidity.

[0045] The following examples are provided for the purpose of illustrating the present invention and are not intended to limit the scope of the invention.

#### EXAMPLES

[0046] Core

<tb><sep>Quantity (mg)<sep>

<tb><sep>Ingredient<sep>Example 1<sep>Example 2

<tb><sep>Intragranular<sep><sep>

<tb><sep>**Gabapentin**<sep>800<sep>800

<tb><sep>Hydroxypropyl cellulose-L (HPC-L)<sep>40<sep>40

<tb><sep>Crospovidone<sep>22<sep>-

<tb><sep>Extragranular

<tb><sep>Crospovidone<sep>22<sep>44

<tb><sep>Corn starch<sep>60<sep>-

<tb><sep>Poloxamer<sep>11<sep>11

<tb><sep>Dicalcium phosphate<sep>68<sep>-

<tb><sep>Mannitol<sep>110<sep>178

<tb><sep>Talc<sep>11<sep>11

<tb><sep>Magnesium stearate<sep>16<sep>16

[0047] Method:



Example 1

[0048] **Gabapentin**, HPC-L (half quantity) and crospovidone are mixed in a rapid mixed granulator and granulated with a HPC-L solution/dispersion in purified water and dried in a fluid bed dryer. The resulting dried granules are mixed with the extragranular excipients, i.e., crospovidone, corn starch, poloxamer, dicalcium phosphate and mannitol, in a low shear blender for 15 minutes. The resulting blend is mixed with talc and magnesium stearate in a low shear blender for 10 minutes and compressed into tablets using appropriate tooling.

Example 2

[0049] **Gabapentin** and HPC-L (half quantity) are mixed in a rapid mixer granulator and granulated with a binder solution (i.e., the solution of the rest of the quantity of HPC-L in purified water) and dried in a fluid bed dryer. The resulting dried granules are mixed with the extragranular excipients, i.e., crospovidone, poloxamer and mannitol, in a low shear blender for 15 minutes. The resulting blend is finally mixed with talc and magnesium stearate in a low shear blender for 10 minutes and compressed into tablets using appropriate tooling.

[0050] The tablets made as per the above examples are coated with a coating having the following composition:

[0051] Coating formula:

Hydroxypropylcellulose: 15 mg

Talc: 15 mg

Purified water: q.s.

[0055] The tablets of Example 2 were subjected to accelerated studies for three months at 40[deg.] C. and 75% relative humidity (RH). The resulting stability, friability and hardness data are shown in Tables 1 and 2.

<tb>TABLE 1

<tb>Stability data of **gabapentin** tablets

<tb>subjected to accelerated studies.

<tb><sep><sep>1 M/40[deg.] C./<sep>2 M/40[deg.] C./<sep>3 M/40[deg.] C./

<tb><sep>Initial<sep>75 % RH<sep>75% RH<sep>75% RH

<tb>**Gabapentin**<sep>99.51<sep>97.02<sep>101.4<sep>99.04

<tb>(% w/w)

<tb>**Gabapentin**<sep>N.D\*<sep>0.027<sep>0.139<sep>0.198

<tb>lactam

<tb>derivative

<tb>(% w/w)

<tb>

\*N.D.-Not detected

[0056]

<tb>TABLE 2

<tb>Friability and Hardness data

<tb><sep><sep>Friability<sep>Hardness

<tb><sep>Tablet<sep>(% w/w)<sep>Range (Kp)

<tb><sep>Uncoated tablets<sep>0.25<sep>16-18

<tb><sep>Coated Tablets (initial)<sep>0.03<sep>20-24

<tb><sep>Coated Tablets (One month<sep>0.00<sep>22-27

<tb><sep>at 40[deg.] C./75% RH)

<tb><sep>Coated Tablets (Two months<sep>0.10<sep>19-24

<tb><sep>at 40[deg.] C./75% RH)

<tb><sep>Coated Tablets (Three months<sep>0.04<sep>20-22

<tb><sep>at 40[deg.] C./75% RH)

[0057] As can be seen from Tables 1 and 2, the tablets have very low levels of the lactam and acceptable friability and hardness values initially and after storage at 40[deg.] C. and 75% relative humidity for up to three months.

[0058] While several particular forms of the inventions have been described, it will be apparent that various modifications and combinations of the inventions detailed in the text can be made without departing from the spirit and scope of the inventions. For example, the **gabapentin** tablets described herein can be used for any approved or unapproved use for which **gabapentin** provides therapeutic benefit. These uses include but are not limited to: (1) as an adjunctive therapy in the treatment of partial seizures with and without secondary generalization in adults with epilepsy; (2) as an anticonvulsant used to control various types of seizures in the treatment of epilepsy; (3) for post poliomyelitis pain and amyotrophic lateral sclerosis; (4) for controlling rapid cycling and mixed bipolar states in people who have not received adequate relief from carbamazepine and/or valproate; (5) treatment for the pain of diabetic neuropathy; (6) reducing the pain from chronic neuropathic pain (e.g., due to damaged nerves) while also reducing sleep disturbances and improving mood and enhancing patients' quality of life; and (7) as a prophylactic agent for patients with migraine headaches. Reports indicate that **gabapentin** also can be used for the treatment of pain (neuropathies, neuralgias, fibromyalgia, chronic, back, headache, migraine), bipolar affective disorder, epilepsy, restless leg, multiple sclerosis, anxiety, and behavior disorders. The **gabapentin** drug products made according to the methods disclosed herein also can be used for these indications and treatments. Further, it is contemplated that any single feature or any combination of optional features of the inventive variations described herein may be specifically excluded from the claimed inventions and be so described as a negative limitation. Accordingly, it is not intended that the inventions be limited,

except as by the appended claims.

---

Data supplied from the *esp@cenet* database - Worldwide



# Effects of anticonvulsant drug gabapentin on the enzymes in metabolic pathways of glutamate and GABA

**Authors:** Goldlust A.; Ti-Zhi S.; Welty D.F.; Taylor C.P.; Oxender D.L.<sup>1</sup>

**Source:** Epilepsy Research, Volume 22, Number 1, September 1995, pp. 1-11(11)

**Publisher:** Elsevier Science

- [next article >](#)
- |
- [View Table of Contents](#)

<mark item

[Article access options](#)

## Abstract:

Gabapentin is a novel anticonvulsant drug. The anticonvulsant mechanism of gabapentin is not known. Based on the amino acid structure of gabapentin we explored its possible effects on glutamate and  $\gamma$ -aminobutyric acid (GABA) metabolism in brain as they may relate to its anticonvulsant mechanisms of action. Gabapentin was tested for its effects on seven enzymes in the metabolic pathways of these two neurotransmitters: alanine aminotransferase (AL-T), aspartate aminotransferase (AS-T), GABA aminotransferase (GABA-T), branched-chain amino acid aminotransferase (BCAA-T), glutamine synthetase (Gln-S), glutaminase (GLNase), and glutamate dehydrogenase (GDH). In the presence of 10 mM gabapentin, only GABA-T, BCAA-T, and GDH activities were affected by this drug. Inhibition of GABA-T by gabapentin was weak (33%). The  $K_i$  values for inhibition of cytosolic and mitochondrial forms of GABA-T (17-20 mM) were much higher than the  $K_m$  values for GABA (1.5-1.9 mM). It is, therefore, unlikely that inhibition of GABA-T by gabapentin is clinically relevant. As with leucine, gabapentin stimulated GDH activity. The GDH activity in rat brain synaptosomes was activated 6-fold and 3.4-fold, respectively, at saturating concentrations (10 mM) of leucine and gabapentin. The half-maximal stimulation by gabapentin was observed at approximately 1.5 mM. Gabapentin is not a substrate of BCAA-T, but it exhibited a potent competitive inhibition of both cytosolic and mitochondrial forms of brain BCAA-T. Inhibition of BCAA-T by this drug was reversible. The  $K_i$  values (0.8-1.4 mM) for inhibition of transamination by gabapentin were close to the apparent  $K_m$  values for the branched-chain amino acids (BCAA) l-leucine, l-isoleucine, and l-valine (0.6-1.2 mM), suggesting that gabapentin may significantly reduce synthesis of glutamate from BCAA in brain by acting on BCAA-T.

**Keywords:** Neurontin;  $\gamma$ -Aminobutyric acid-aminotransferase; Glutamate dehydrogenase; Branched-chain amino acid aminotransferase

**Language:** English

**Document Type:** Research article

**DOI:** 10.1016/0920-1211(95)00028-9

**Affiliations: 1:** Department of Biotechnology, Parke-Davis Pharmaceutical Research, Division of Warner-Lambert Co., 2800 Plymouth Rd., Ann Arbor, MI 48105, USA

Access to this article is through a website external to IngentaConnect

You may be required to register and activate access/purchase before you can obtain the full text.

[Proceed](#)

- [next article >](#)
- |
- [View Table of Contents](#)

[Back to top](#)

© 2006 Ingenta    [Terms and Conditions](#) | [Privacy Policy](#)